Purification and Properties of Thermostable Alkaline Protease from *Bacillus subtilis* K-30

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B. subtilis K-30 isolated from soil sample collected from detergent industry produced alkaline protease. Thermostable extracellular alkaline protease (EC 3.4.21.14), suitable for use in detergents, was purified by ion exchange chromatography from *Bacillus subtilis* K-30 and its properties were examined. The purified enzyme had a molecular weight of 29kDa as determined by SDS-PAGE. The optimal pH and temperature of the purified enzyme was found to be 10 and 55°C, respectively. It was capable of hydrolyzing many soluble and insoluble protein substrates. Casein was the best substrate among the substrates tested (BSA, egg albumin, gelatin). The enzyme activity was inhibited by Cu²⁺, Hg²⁺ and Zn²⁺ at 5mM whereas Ca²⁺, Mg²⁺, and Mn²⁺ had stimulatory effect. The protease was 100% inhibited by PMSF, indicating serine protease activity. The enzyme retained more than 40 percent activity after 30 min incubation at 60°C in the presence of detergents such as Henko and Vim ultra, indicating its suitability in detergent industry.

Key words: Bacillus subtilis K-30, Alkaline protease, Thermostable.

Proteases (peptidyl-peptide hydrolases) are one of the most important classes of enzymes and expressed throughout the animal and plant kingdoms as well as in viruses and bacteria (Vikash Dubey *et al.*, 2007). Proteases have been divided into six mechanistic classes by the International Union of Biochemistry. These include the cysteine, serine, aspartic, mettaloprotease, threonine and unknown type (Enzyme nomenclature, 1992). The threonine protease is the most recently discovered type (Seemuller *et al.*, 1995). The protease family has drawn special attention as these constitute one of the most important groups of industrial enzymes accounting for about 65% of the total worldwide enzyme sales (Reddy *et al.*, 2008; Nascimento and Martins, 2004; Beg and Gupta, 2003; Ellaiah *et al.*, 2003, Ito *et al.*, 1998).

Proteases have wide range of applications in detergent, pharmaceutical, leather, food and silk industries. Although a variety of proteolytic fungi and bacteria are known, only few

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provide high activities with commercial success (Berla and Suseela, 2002). Commercial proteases are mostly from various sources and it was reported that about 35% of the total microbial enzymes used in detergent industry are the proteases from bacteria (Ferrero et al., 1996). Bacillus spp. are extensively exploited for protease production (Karl Heinz, 2004). The alkaline proteases from these species represent the lead molecules for the subtilisins. Members of subtilisin super family of proteases have now been identified with different functions, in all living organisms (Siezen and Leunissen, 1997). Thermophiles such as Thermus aquaticus (Gabriela et al., 2003), Bacillus licheniformis (Ferrero et al., 1996), Bacillus stearothermophilus (Boonayanan et al., 2000), Bacillus pumilus (Kumar, 1999), Thermoanerobacter vonsieensis (Hyenung et al., 2002), Bacillus thermoruber (Manachini et al., 1988) and Bacillus brevis (Banerjee et al., 1999) have been studied for their capability to produce thermostable proteases.

Thermostable proteases are advantageous as they are highly stable at elevated temperatures; reduce risk of contamination by other organisms, ideal models for studying thermal stability of protein, elucidation of mechanisms involved in thermostability of enzymes (Helmann, 1995). Most enzymes used are mainly derived from mesophilic sources that work in narrow ranges of pH and moderate temperatures. However, alkaline proteases exhibit optimum activity and stability at high pH, temperatures, in presence of surfactants and chaotropic agents, which make them suitable candidates for industrial applications (Kumar and Takagi, 1999).

In the present study, we report the purification and some properties of thermostable alkaline protease from *Bacillus subtilis* K-30 originally isolated from soil (Naidu and Devi, 2005) using rice bran.

MATERIAL AND METHOD

All the chemicals and microbiological media used in the present study were of analytical grade.

Methods

Production of protease from B. subtilis K-30

Bacillus subtilis K-30 was first isolated in the laboratory from soil sample collected from detergent industry, Nellore, Andhra Pradesh, India (Naidu and Devi, 2005). Production of protease from *B. subtilis* K-30 was carried out in a medium containing (g/L) glucose, 1.0; casein, 0.5; yeast extract, 0. 5; KH₂PO₄, 0.1; and Na₂CO₃, 1.0 and maintained at 37°C for 48 hours in a continuous shaker incubator (120 rpm). After the completion of fermentation, the whole fermentation broth was centrifuged at 10,000 rpm at 4°C, and the clear supernatant was recovered. The crude enzyme was assayed for proteolytic activity and used for further studies.

Purification of the protease

Crude enzyme was precipitated with ammonium sulfate at 60% saturation. All subsequent steps were carried out at 4°C. The protein pellet obtained after saturation with ammonium sulfate was dissolved in 0.1M Tris-HCl buffer and loaded onto column of Sephadex G-100 (1.5X24 cm) (Sigma-Aldrich, St. Louis, MO) equilibrated with Tris-HCl buffer, pH 7.8. The column was eluted at a flow rate of 50 ml/h. From the elution profile, it was observed that the protease was eluted as a single peak of caseinase activity. The collected precipitates were dissolved in 10ml of 0.1M pH 7.8 Tris-HCl buffer. The solution was applied on DEAE column (1.6X40 cm) and eluted with 20mM pH 7.8 Tris-HCl buffer at a flow rate of 0.5ml/min. The protein content was determined using the method previously described (Lowry et al., 1951) with bovine serum albumin (BSA) as standard.

Enzyme assay

Alkaline protease activity was determined by a modification procedure based on the method of casein digestion (Tsuchida *et al.*, 1986). The reaction mixture consists of 0.5ml of purified enzyme, 0.5ml Tris-HCl buffer (50mM, pH 8) and 1ml of 1% casein in Tris-HCl buffer of same pH. The mixture was incubated for 30 min at 55°C and the reaction was stopped by the addition of 4ml of 5% Trichloroacetic acid (TCA). After 60 min, the solution is centrifuged at 4°C for 15 min at 3000 rpm. 1 ml of supernatant was mixed with 5 ml of 0.4M Na₂CO₃ followed by the addition of 0.5 ml phenol reagent and absorbance

was read at 660nm. One protease Unit is defined as the amount of enzyme that releases 1μ mol of tyrosine per minute under the above said condition.

Polyacrylamide Gel Electrophoresis

After DEAE column chromatography, the fractions¹⁹⁻²³ showing the highest specific activity were dialyzed, lyophilized and subjected to sodium dodecyl sulphate polyacrylamide gel electrophoresis (SDS-PAGE). SDS-PAGE was carried out at 12% (w/v) resolving gel and 5%(w/ v) stacking gel according to the method of Laemmili 1971). The following proteins were used as protein standards: BSA (66 kDa), ovalbumin (45 kDa), glyceraldehyde-3-P-dehydrogenase (36 kDa), carbonic anhydrase (29 kDa), trypsinogen (24 kDa), soybean trypsin inhibitor (20kDa), and lactalbumin (14.2 kDa).

Effect of pH on purified enzyme activity and stability

The activity of crude and purified protease was measured at different pH values in the presence and absence of 10mM CaCl₂. The pH was adjusted using the following buffers (0.05M): phosphate (pH 5.0-7.0), Tris-HCl (pH 8.0) and glycine –NaOH (pH9.0-12.0). Reaction mixtures were incubated at 55°C for 30 minutes, and the activity of enzyme was measured. The purified enzyme was diluted in different buffers (pH 5.0-12.0) and incubated at 55°C for 2 and 20 h for enzyme stability. The relative activity at each exposure was measured.

Effect of temperature on enzyme activity and stability

The activity of the crude and purified enzyme was determined by incubating the reaction mixture at different temperatures ranging from 30° C to 90°C in the presence and absence of 10mM CaCl₂ for 30 minutes. Thermal stability was studied as described (Huang *et al.*, 2006) by preincubating the purified enzyme without substrate at 60, 70, and 80°C from 0-180 min, before determination of proteolytic activity on casein at 1%.

Effect of inhibitors on protease activity

The effect of various inhibitors (5mM) such as serine inhibitors phenyl methyl sulfonyl fluoride (PMSF) and diisopropyl fluro phosphate (DFP), cysteine inhibitors p-chloro mercuric benzoate (p CMB) and β -mercaptoethanol

(β -ME), iodoacetate and a chelator of divalent cations ethylene diamine tetraacetic acid (EDTA) was determined by preincubation with the enzyme solution for 30 minutes at 60°C before the addition of substrate. The relative protease activity was measured.

Effect of various metal ions on protease activity

The effects of metal ions $(Ca^{2+}, Mg^{2+}, Al^{3+}, Co^{2+}, Cd^{2+}, Fe^{3+}, Na^{2+}, Zn^{2+}, Hg^{2+}, and Cu^{2+}$ (5mM) were investigated by adding them to the reaction mixture. Relative protease activities were measured.

Substrate specificity

Protease activity with various protein substrates including bovine serum albumin, casein, egg albumin and gelatin was assayed by mixing 100ng of enzyme and 200µl of assay buffer containing the protein substrates (2mg/ml). After incubation at 60°C for 30 minutes, the reaction was stopped by adding 4ml of 5% TCA and allowed to stand for 10 min. The undigested protein was removed by centrifugation and peptides released were assayed. The specific protease activity towards casein was taken as a control.

Detergent stability

The compatibility of *B. subtilis* K-30 protease with local laundry detergents was studied in the presence of 10mM CaCl₂ and 1M glycine. Detergents used were Nirma (Nirma chemical, India); Henko (Henkel spic, India); Surf, Vim, wheel, Rin (Hindustan Lever Ltd, India) and Ariel (Procter and Gamble, India). The detergents were diluted in distilled water (0.7% w/v) and incubated with purified protease for 3 h at 60°C, and the residual activity was determined. The enzyme activity of a control sample (without any detergent) was taken as 100%.

RESULTS

Purification of alkaline protease

The results on the purification of the protease from *B. subtilis* K-30 are summarized in Table 1. The protein pellet obtained after 60% saturation with ammonium sulphate was dissolved in 0.1M Tris-HCl buffer and loaded onto a column of Sephadex G-100 (1.5X24 cm) equilibrated with Tris-HCl buffer, pH 7.8. Fractions (13-20) with high protease activities were pooled, and used for

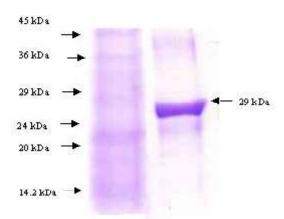


Fig. 1. SDS-PAGE of thermostable alkaline protease from *B. subtilis* K-30. Lanel molecular mass markers. The molecular mass values of the markers are ovalbumin (45 kDa), glyceraldehydes -3phosphate dehydrogenase (36 kDa), carbonic anhydrase (29 kDa), trypsinogen (24 kDa), soybeantrpsin inhibitor (20 kDa), and lactalbumin (14.2 kDa). Lane 2, purified protease

further studies. Overall 94.33 fold purification, 14.15 U mg⁻¹protein increase in specific activity was achieved with 10.63% recovery. The molecular weight of purified enzyme as determined by SDS-PAGE was found to be 29 kDa (Fig.1). The appearance of single band on SDS– PAGE suggested the enzyme is purified to homogenity. Molecular weight in the range of 28 -32 kDa has also been reported other *Bacillus* sp. Strain OK 3a.1.b-18 and BAH-101 (Coolbear *et al.*, 1981; Fuziwara *et al.*, 1993).

Effect of pH on protease activity

The pH optimum of the purified enzyme was found to be at pH 10.0 (Fig. 2). These findings are in accordance with several earlier reports showing pH optima of 10-10.5 for protease from *Bacillus* sp, *Thermus aquaticus*, *Vibrio metscnikovii* (Matsuzawa *et al.*, 1988; Betzel *et al.*, 1992; Kwon *et al.*, 1994). The important

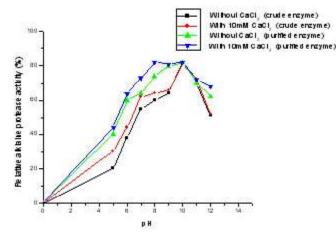


Fig. 2. Effect of pH on alkaline protease activity in presence and absence of 10mM CaCl, at 45°C

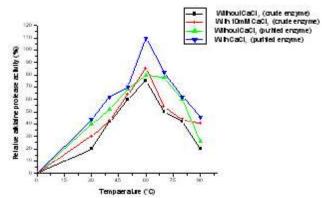


Fig. 3. Effect of temperature on alkaline protease activity in presence and absence of 10mM CaCl,

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detergent enzymes subtilisin NOVO or BPN (Horikoshi, 1990) showed maximum activity at pH 10.5. The enzyme was almost 100% stable over a broad pH range of pH 8-10 for 4 h incubation and a pH range of 8-10 for 20 h at 40°C.

Effect of inhibitors and chelators on protease activity

The effect of different inhibitors on the enzyme activity of the purified protease was studied (Table 2). The enzyme was completely inhibited by PMSF (at 5mM concentration) suggesting that protease is serine protease, while

Purification steps	Volume (ml)	Total activity (U ml ⁻¹)	Total protein(mg)	Specific Activity (U mg ⁻¹ protein)	Recovery (%)	Fold purification
Culturefiltrate	500	235	1600.00	0.15	100	-
$(NH_4)_2 SO_4 fraction$	25	107	58.00	1.90	45.53	12.66
DEAE-cellulose	165	85	14.00	6.07	36.17	40.46
SephadexG-100	38	25	1.65	14.15	10.63	94.33

Table 1. Summary of Purification steps of alkaline protease from Bacillus subtilis K-30

Table 2. Effect of inhibitors on protease activity
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Inhibitor/Chelator % (5mM)	Relative enzyme activity		
Control	100		
PMSF	00		
DFP	04		
pCMB	85		
βΜΕ	83		
Iodoacetate	92		
EDTA	102		

Table 4. Alkaline protease activity against different natural substrates

Substrate	Relative Activity%		
Casein	100		
BSA	40		
Egg albumin	23		
Gelatin	40		
Collagen	0		
Keratin	0		
Azocasein	35		

Table 3. Effect of metal ions on protease activity

Metal ions (5mM)	Residual alkaline protease Activity (%)		
Control	100		
Ca^{2+}	128		
Mg^{2+}	115		
Mn ²⁺	102		
Zn^{2+}	96		
Cu^{2+}	94		
Hg^{2+}	90		
Co ²⁺	92		
Na^{2+}	95		
Cd^{2+}	90		
Al ³⁺	93		
EDTA	99		

Table 5. Compatibility 0f Alkaline Proteaseactivity from *Bacillus subtilis* K-30 withcommercial detergents in the presenceof 10mM $CaCl_2$ and glycine

Detergent	% retention of enzyme activity after minutes				
Control	10	30	50	60	
Wheel	100	100	100	100	
Nirma	55	37	25	0	
Surf	43	26	0	0	
Ariel	20	0	0	0	
Henko	60	40	25	0	
Rin	65	55	25	0	
Vim(ultra)	45	25	0	0	
	84	62	35	25	

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DFP exhibited 96% inhibition. In this regard, PMSF is known to sulphonate the essential serine residue in the active site of the protease resulting in a total loss of enzyme activity (Gold and Fahrney, 1964). Partial or no inhibitory effect on the protease activity was observed with other protease inhibitors such as EDTA. However, metal ions such as Ca²⁺, Mg²⁺, Mn²⁺(Table 3) increased and stabilized the protease activity, confirming that these cations take part in the stabilization of the protease structure and play important role in protection against thermal denaturation (Paliwal *et al.*, 1994).

Properties of protease of *B. subtilis* K-30 Effect of temperature on protease activity

The optimum temperature for purified enzyme was found to be 55° C- 60° C. The enzyme activity of the crude and purified enzyme was determined at different temperatures ranging from 30° C- 90° C in the absence and presence of 10mM CaCl₂ (Fig. 3). Optimum temperature of 55° C was recorded earlier for an alkaline protease from *Bacillus cereus* MCM B-326 (Nilegaonkar *et al.*, 2007; Prakash et al., 2005), *Bacillus* sp. (Gupta *et al.*, 2005), *B. licheniformis* RP1 (Sellami-Kamoun *et al.*, 2006) *B. stearothermophilus* AP-4 (Fuziwara and Yamamato, 1987; Dhandapani and Vijayaraghavan, 1994).

Substrate specificity and detergent stability of alkaline protease

The ability to hydrolyze several protein substrates is an important criterion of protease potency (Grebeshova, 1999). Table 4 compares the digestive capability of this protease. The protease showed a high level of hydrolytic activity against casein and poor to moderate hydrolysis of BSA, gelatin and egg albumin, although the hydrolysis was not observed with substrates such as collagen and keratin.

Besides having stability at high pH & temperature, a good detergent protease is expected to be stable in presence of commercial detergents. The K-30 protease showed excellent stability and compatibility with locally available detergents (Table 5) at 60°C in presence of CaCl₂ and glycine as stabilizers. The enzyme retained more than 40 per cent activity after 30 min incubation at 60°C in the presence of detergents such as Henko and Vim ultra, indicating its suitability in detergent industry.

DISCUSSION

A highly thermostable alkaline protease from Bacillus subtilis K-30 was purified with Sephadex G-100 column chromatography. The molecular weight was found to be around 29kDa by SDS-PAGE. The optimal pH and optimal temperature of the protease were at pH 10 and 55°C, respectively. The protease activity was inhibited by PMSF, suggesting that it is a serine protease. The enzyme activity was markedly inhibited by Cu²⁺, Hg²⁺ and Zn²⁺ at 5mM whereas Ca²⁺, Mg²⁺, and Mn²⁺ had stimulatory effect. Considering the different characteristics of high pH and thermostability, compatibility with local and commercial detergents indicate the possibility of use in detergent industry and various other biotechnological applications

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