

Synthesis, Characterization of TiO₂ Doped Nanofibres and Investigation on their Antimicrobial Property

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Abstract

Contaminants in air and water contribute largely to the occurrence of a number of diseases and health problems. TiO₂ has been used as a photo-catalyst in various fields such as water purification, air-filters, disinfection and sterilization. The present study is focused on the synthesis of TiO₂ and TiO₂ doped nanofibers to achieve highly efficient air filters and to obtain increased surface area to volume ratio and optimize concentration of the Ag and Zn in the main matrix of TiO₂ to get the better inhibition of both bacterial and fungal species. The synthesized nanofibers were further characterized by employing the standard techniques such as FTIR, SEM and X-RD for Chemical, Morphological and for Structural studies, respectively. The antibacterial of bacterial species: *Bacillus subtilis* (BS), *Staphylococcus epidermidis* (SE), *Staphylococcus aureus* (SA) and *Staphylococcus faecalis* (SF) and fungal species *Candida tropicalis* (CT), *Candida parapsilosis* (CP), *Candida albicans* (CA) and *Candida glabrata* (CG) were evaluated.

Keywords: Nanofibers, TiO₂, Photocatalyst, Antimicrobial activity, SEM, XRD, FTIR, Microfibrus.

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INTRODUCTION

Preparation of nanofibers in recent years has been carried out by various processing techniques such as pattern synthesis, sketching, separating into two phases, electrospinning and self-accumulation¹. Though the principle of electrospinning has been used since 1930s, the term was first used in around 1994. Many polymers will be dissolved in solvents and few heated into melts, from thus ultra fine fibers have been spun using this technique²⁻⁴. The Traditional micro sized fibers have applications especially in engineering fibers such as carbon, glass, and Kevlar fibers, is to be used as reinforcements in composite developments⁸. Air pollution poses to be a major threat to the environment and human health. Organic compounds, microorganisms, pollutants, allergens, and rest in air and water have proven to be highly toxic in nature. These compounds are smaller than 1 micrometer in diameter, and can thus easily enter our bodies via air and water. Using nanofibers of smaller diameter in filters enables the increase in the efficiency of its filtration application³¹. The bactericidal activity of TiO₂ as a photocatalyst was discovered by Matsunaga *et al.*¹⁴ in the year 1985. TiO₂ is known for its photo catalytic inactivation of various organisms like bacteria¹⁵⁻¹⁸, algae¹⁹ and bacterial and fungal spores²⁰⁻²². Since then TiO₂ has been used as a photocatalyst in various fields such as water purification, air filters, disinfection, and sterilization.

In each of the studies the method of preparation of the TiO₂ nanofibers varies. Titanium (IV) tetra isopropoxide and Titanium (IV) tetra butoxide are the commonly used precursors while Polyvinylpyrrolidone or Dimethyl formamide have been used as a solvent in preparation of spinning solution. Table 1 gives a brief review of literature including the chemicals used in the studies.

Studies have also shown nanofibrous membranes to have high efficiency in filter and pre filter applications. The accumulation or capture of sub-micron particles on nanofibers media has been observed to be relatively high while compared to that of standard media (Schaeffer and Olson 1998; Schaeffer *et al.* 1998). The best way of increasing efficiency of filters is by considering the parameters such as Porosity, Fiber Diameter^{32,33}, Pore Size, Surface Area, Anti-microbial activity of

nanofibers³⁴.

Thus research has been carried out over the years to modify this property of TiO₂ by doping TiO₂ with a suitable dopant. Nitrogen, Sulphur and carbon have been used as dopants to obtain successful visible range photocatalytic activity. A visible-light activated PdO/TiON photocatalyst was formed on combining Pd ion and nitrogen. This has shown remarkable photocatalytic activities on a number of organic¹⁷ and microbiological species.

On microfibrinous scaffolds the cells are attached on one side due to its polarized nature and the other side was exposed to the air/media. Where as in nanofibrous scaffolds, it is likely that cells are more naturally guarded. The mechanism of antimicrobial activity of silver is partially understood. It has been proposed that the binding of silver ions to multiple sites may contribute to structural and functional changes in the cell leading to disruption of the cell²⁶. The inactivation of enzymes due to the binding of silver ions with the bifunctional groups on the cell surface is another proposed theory of bactericidal activity²⁷. Nanofibers of cellulose acetate, poly acrylonitrile and polyvinylchloride with 5% silver nitrate exhibited positive results when tested for antibacterial properties²⁶. PVA- Ag nanofibers obtained by electrospinning was studied and showed strong antibacterial properties²⁷. Studies showed that silver/polyrhodamine nanofiber has the enhanced antimicrobial efficacy than silver sulfadiazine nanofiber²⁸. Silver nanoparticles on polyurethane nanofibers showed high bactericidal effects²⁹. The present study has been focussed on making of nanofibers of TiO₂; Ag and TiO₂; Zn for effective treatment of the air filtration and study its antimicrobial activity.

The Nanofibers were prepared by preparing stock solution of TiO₂ as base matrix, Polyvinylpyrrolidene stock as a binder and to enhance the activity silver, tin and zinc in different concentrations as dopants in the main matrix by using Sol-gel process. The prepared viscous solution has been loaded into the 5ml of syringe in the Electro spin cabinet, and required high voltage applied to draw the desired nanofibers with specific diameter. The distance between the collector and syringe tip has been adjusted to get the optimized nanofibers diameter. The nanofibers were drawn on the Al foil and

collected using a drum-collector. The synthesized nanofibers were tested for antimicrobial activity using Drop test method for four fungal and for four Bacterial species, by standardizing the solution with Mc Farland standard solution. The synthesized nanofibers were further characterized by employing the standard techniques such as FTIR, SEM and X-RD for Chemical, Morphological and for Structural studies, respectively. From the studies, it is evident that increasing the dopant concentration to metal oxide TiO_2 there by a decrease in the microbial colony formation units per hour. We have optimized the concentration of the Ag and Zn in the main matrix of TiO_2 to get the better inhibition of both bacterial and fungal species. The optimized composition is further tested for air filtration application.

EXPERIMENTAL METHODOLOGY

Optimization of the Polymer Nanofibers (Polyvinylpyrrolidene Polymer)

Different weight % of PVP was added into a beaker containing appropriate amount of solvent and stirred to obtain a viscous solution. The solution obtained was transferred into a 5ml syringe and electrospun by choosing appropriate flow rate, distance and voltage. The nanofibers collected on the aluminum foil were characterized by SEM to observe the morphology.

Preparation of TiO_2 Stock Solution

Titanium isopropoxide solution was prepared by dissolving titanium isopropoxide (Aldrich, 99.98% purity) in 2-methoxy ethanol (99.98%) solvent. Di acetyl-ketone was added as the complexing and chelating agent. 1.2 g of stabilizing agent based on cationic surfactant was dissolved in 20 ml of 2-methoxy ethanol was added drop wise to the titanium isopropoxide solution under vigorous stirring, the stirring being carried out for about 6 hours. The molar ratio of titanium isopropoxide, 2-methoxy ethanol and Di acetyl-ketone was 2.0:10.0:0.5. To increase the rate of reaction and hydrolysis of the solution calculated quantities of HNO_3 and deionised water were each added drop wise. For 3h at 70°C the solution was refluxed and then cooled to room temperature. Filtration of the contents of the solution was carried out over Whatman filter paper, to remove any particulates formed during the reaction. The

resulting TiO_2 solution was used as the stock solution for synthesizing nanofibers of TiO_2 ⁴⁸.

Optimization of TiO_2 with Polymer Binder Polypyrrolidene Polymer

Different weight % of PVP was added into a beaker containing appropriate amount of solvent and stirred to obtain a viscous solution. The required amount of TiO_2 stock solution is also added and the contents are stirred to obtain a homogeneous solution. The solution obtained was transferred into a 5ml syringe and electrospun by choosing appropriate flow rate, distance and voltage. The nanofibers collected on the aluminum foil were characterized by SEM to observe the morphology.

Optimization of TiO_2 : Ag with Polymer Binder Polypyrrolidene Polymer

Different weight % of PVP was added into a beaker containing appropriate amount of solvent and stirred to obtain a viscous solution. The required amount of TiO_2 stock solution is also added and the contents are stirred to obtain a homogeneous solution. Calculated quantity of the silver nitrate is taken in 50 ml beaker and to it 2-methoxy ethanol solvent is added to make a homogenous solution. Required quantity of the obtained solution is added to the TiO_2 – PVP solution and the contents are stirred. The final solution obtained was transferred into a 5ml syringe and electrospun by choosing appropriate flow rate, distance and voltage. The nanofibers collected on the aluminum foil were characterized by SEM to observe the morphology.

Optimization of TiO_2 : Zn with Polymer Binder Polypyrrolidene Polymer

Different weight % of PVP was added into a beaker containing appropriate amount of solvent and stirred to obtain a viscous solution. The required amount of TiO_2 stock solution is also added and the contents are stirred to obtain a homogeneous solution. Calculated quantity of the zinc nitrate is taken in 50 ml beaker and to it 2-methoxy ethanol solvent is added to make a homogenous solution. Required quantity of the obtained solution is added to the TiO_2 – PVP solution and the contents are stirred. The final solution obtained was transferred into a 5ml syringe and electrospun by choosing appropriate flow rate, distance and voltage. The nanofibers

collected on the aluminum foil were characterized by SEM to observe the morphology.

Antimicrobial Tests using Drop Test Method

(a) Antibacterial Activity Test

Nutrient agar medium was prepared containing peptone (5.0 g), beef extract (3.0 g), and sodium chloride (NaCl) (5.0 g) in 1000 ml distilled water. The pH was adjusted to 7.0 and agar (15.0g) was added to the solution. In 15ml aliquots the media was sterilized at a pressure of 15 PSI for 15 min. The nutrient agar media was allowed to solidify by transferring the media into sterilized petri dishes and placing the petri dishes in a laminar air flow unit. Bacterial suspensions (for each species tested) are grown in nutrient broth for 24 h. This is standardized to a 0.5 McFarland standard solution. 100 μ l of the bacterial suspensions (for each species tested), were added onto plates into which TiO₂ or TiO₂: Ag nanofibres were transferred. These plates were kept in a UV chamber for 4 h. At 1 h intervals, exposed plates were washed with sterile saline or distilled water into prepared agar plates. To enable equal distribution of the organisms the resulting culture was spread with a glass spreader. Plates were incubated at 37°C for 24 hrs and colony forming units (CFU) were counted. A positive and negative control was prepared at the same time^{49,50}.

(b) Antifungal Activity Test

Sabouraud agar was prepared containing 40 g/L dextrose, 10 g/L peptone, 20 g/L agar dissolved in 1000 ml distilled water. The pH was adjusted to 5.6. In 15ml aliquots the media was sterilized at a pressure of 15 PSI for 15 min. The agar media was allowed to solidify by transferring the media into sterilized petri dishes and placing the petri dishes in a laminar air flow unit. 24 hours fungal suspension was grown in Sabouraud broth and standardized by comparison with a 0.5 McFarland standard solution. 100 μ L of the fungal suspensions, were added onto plates into which TiO₂ or TiO₂: Ag nanofibres were transferred. These plates were kept in a UV chamber for four hours. At 1 h intervals, exposed plates were washed with sterile saline or distilled water into prepared agar plates. To enable equal distribution of the organisms the resulting culture was spread with a glass spreader. Plates were incubated at 37°C for 24 hour and colony forming units (CFU) were counted. A positive and negative control was

prepared at the same time^{49,50}. Fig. 1 shows the flowchart of antimicrobial testing.

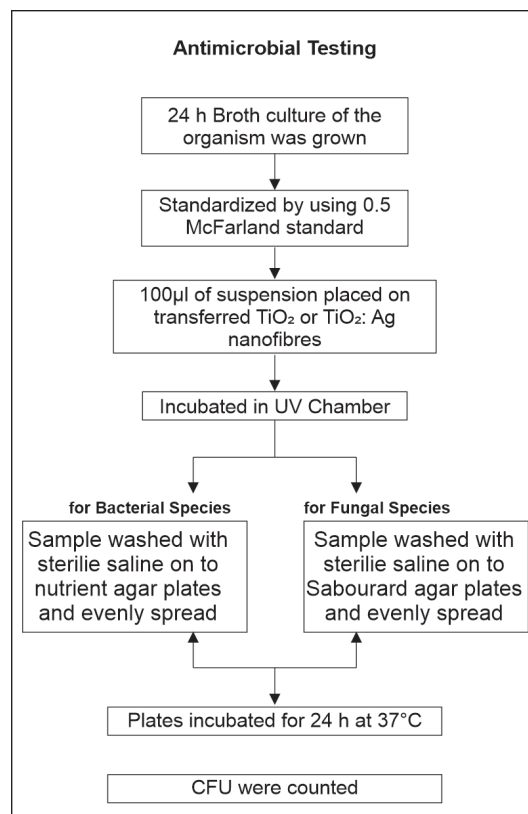


Fig. 1. Flowchart of antimicrobial testing

RESULTS AND DISCUSSIONS BY DROP TEST METHOD

Antimicrobial Tests for TiO₂: Ag concentration

Anti bacterial Testing

The antibacterial testing was carried out for the following bacterial species: *Bacillus subtilis* (BS), *Staphylococcus epidermidis* (SE), *Staphylococcus aureus* (SA) and *Staphylococcus faecalis* (SF). Fig. 2 a and b is showing the results of *Bacillus subtilis* (BS) for TiO₂: Ag. Fig. 2 c and d is showing the results of *Staphylococcus epidermidis* (SE) for TiO₂: Ag.

Each tested bacterial species showed vulnerability to pure TiO₂ and TiO₂ – Ag nanofibres as shown in the figures. Fig. 2 a) and b) indicates antibacterial images of TiO₂ +Ag at different concentrations on *Staphylococcus aureus* a) 0.2% Ag; b) 0.4% Ag. Fig. 2 b) and c) indicates antibacterial images of TiO₂ +Ag at varying

concentrations on *Staphylococcus aureus* c) 0.6% Ag; d) 0.8% Ag. There is an increase in susceptibility observed with increase in the concentration of Ag. At 0.8% Ag concentration *Staphylococcus aureus* shows drastic reduction of number of colonies

formed with increase in Ag concentration and complete removal of organisms after 4 hours of exposure to UV. The other species show a gradual reduction in number of viable organisms with an increase in Ag concentration.

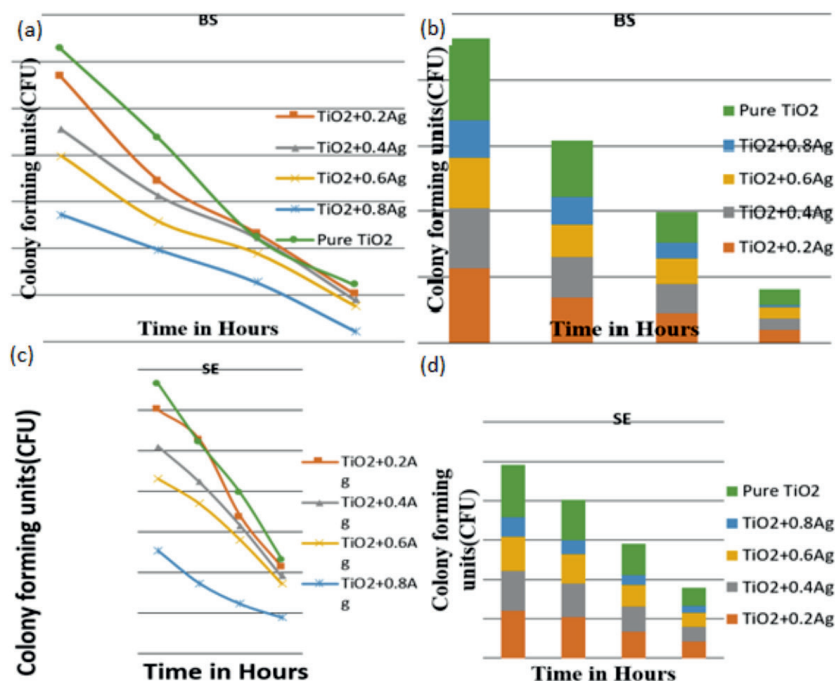


Fig. 2. (a) & (b): *Bacillus subtilis* (BS) for TiO₂: Ag (a) & (b): *Staphylococcus epidermidis* (SE) for TiO₂: Ag

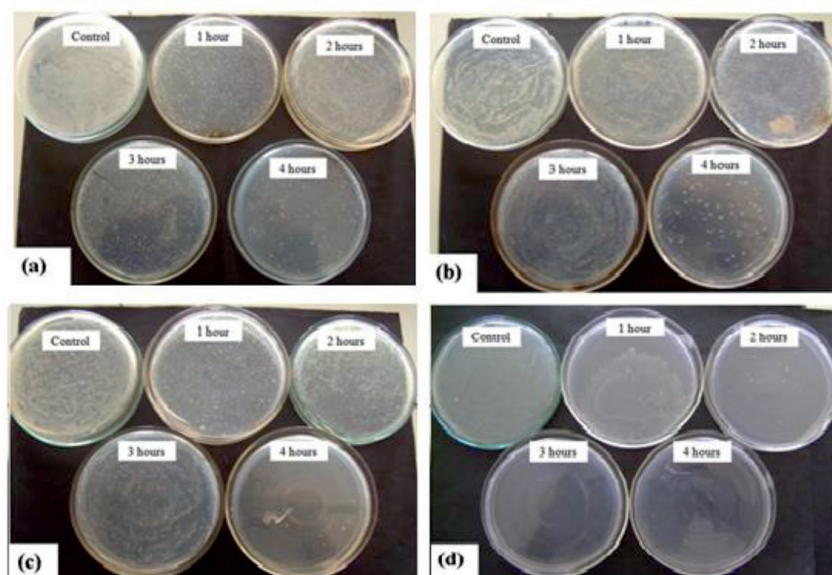


Fig. 3. Antibacterial Images of TiO₂ + Ag at Different Concentrations on *Staphylococcus aureus* a) 0.2% Ag; b) 0.4% Ag c) 0.6% Ag; d) 0.8% Ag

RESULTS AND DISCUSSION

SEM Analysis

The morphology of the obtained nanofibers was analyzed using SEM.

PVP Nanofibers

SEM images of PVP grown at distance 12cm and applied voltage between needle and substrate 12kV.

a) 15wt % PVP in ethanol deposited on Al foil at room temperature.

Fig. 4 a) indicates SEM image of 15% PVP Nanofibers under low magnification. Fig. 4 b) indicates SEM image of 15% PVP Nanofibers under high magnification.

b) 15wt % PVP in ethanol deposited on Al foil annealed at 200°C for 1 h in air ambient

Fig. 4 c) indicates SEM image of 15% PVP Nanofibers (200°C) under low magnification. Fig. 4 d) indicates SEM image of 15% PVP Nanofibers (200°C) under high magnification.

c) 15wt % PVP in ethanol deposited on Al

foil annealed at 400°C for 1 h in air ambient

Fig. 4 e) indicates SEM image of 15% PVP Nanofibers (400°C) under low magnification. Fig. 5 f) indicates SEM image of 15% PVP Nanofibers (400°C) under high magnification.

At room temperature the nanofibers formed have an average diameter of 110.4nm diameter. On annealing at 200°C and 400°C for an hour the nanofibers get disintegrated. Hence for further experimentation 15% PVP in ethanol is taken as the optimized composition for PVP.

TiO₂ – PVP Nanofibers

SEM images of TiO₂: PVP grown at distance 12cm and applied voltage between needle and substrate 12kV.

a) TiO₂: PVP with 5wt % TiO₂ in ethanol deposited on Al foil at room temperature.

Fig. 5 a) indicates SEM image of 5% TiO₂: PVP Nanofibers under low magnification. Fig. 5 b) indicates SEM image of 5% TiO₂: PVP Nanofibers under high magnification.

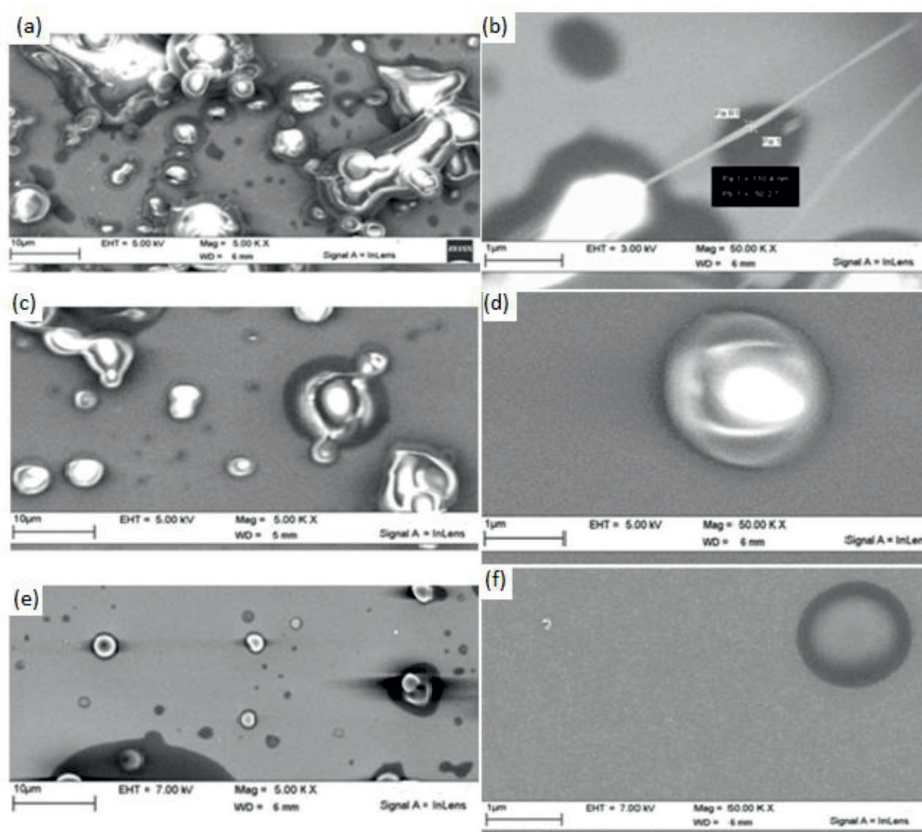


Fig. 4. SEM image of 15% PVP Nanofibers: a) Low Magnification, b) High Magnification, c) Low Magnification (200°C), d) High Magnification (200°C), e) Low Magnification (400°C), f) High Magnification (400°C)

b) TiO_2 : PVP with 10wt % TiO_2 in ethanol deposited on Al foil at room temperature

Fig. 5c) indicates SEM image of 10% TiO_2 : PVP Nanofibers under low magnification.

Fig. 5d) indicates SEM image of 10% TiO_2 : PVP Nanofibers under high magnification.

c) TiO_2 : PVP with 15wt % TiO_2 in ethanol deposited on Al foil at room temperature.

Fig. 5 e) indicates SEM image of 15% TiO_2 : PVP Nanofibers under low magnification. Fig. 5 f) indicates SEM image of 15% TiO_2 : PVP Nanofibers under high magnification.

TiO_2 : Ag Nanofibers

SEM images of TiO_2 : PVP: Ag grown at a distance of 10cm and applied voltage between needle and substrate 12kV.

a) 15wt% PVP in ethanol + 10wt% TiO_2 + 0.2 wt% of Ag.

Fig. 6 a) indicates SEM image of 15% PVP: 10% TiO_2 : 0.2% Ag Nanofibers under low magnification Fig. 6 b) indicates SEM image of 15% PVP: 10% TiO_2 : 0.2% Ag Nanofibers under high magnification.

b) 15wt% PVP in ethanol + 10wt% TiO_2 +

0.4 wt% of Ag.

Fig. 6 c) indicates SEM image of 15% PVP: 10% TiO_2 : 0.4% Ag Nanofibers under low magnification. Fig. 6 d) indicates SEM image of 15% PVP: 10% TiO_2 : 0.4% Ag Nanofibers under high magnification.

c) 15wt% PVP in ethanol + 10wt% TiO_2 + 0.6 wt% of Ag

Fig. 6 e) indicates SEM image of 15% PVP: 10% TiO_2 : 0.6% Ag Nanofibers under low magnification. Fig. 6 f) indicates SEM image of 15% PVP: 10% TiO_2 : 0.6% Ag Nanofibers under high magnification.

d) 15wt% PVP in ethanol + 10wt% TiO_2 + 0.8 wt% of Ag.

Fig. 6 g) indicates SEM image of 15% PVP: 10% TiO_2 : 0.8% Ag Nanofibers under low magnification. Fig. 6 h) indicates SEM image of 15% PVP: 10% TiO_2 : 0.8% Ag Nanofibers under high magnification.

The nanofibers were obtained on varying the concentrations of silver as an additive to enhance the photocatalytic activity of base TiO_2 . Calculated quantity of the silver nitrate is taken in

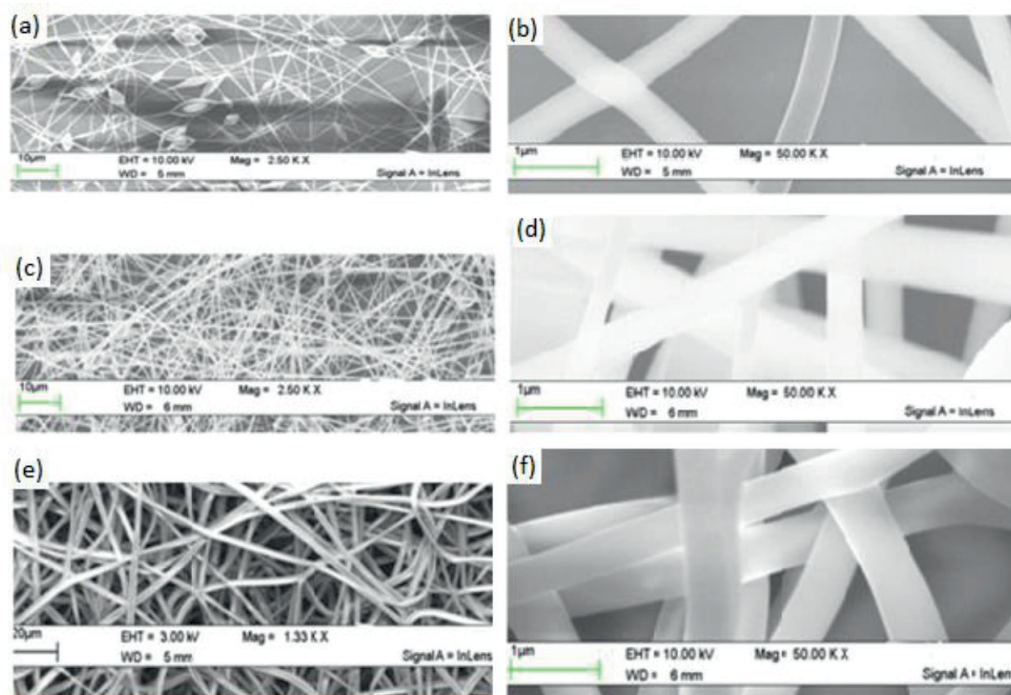


Fig. 5. SEM image of TiO_2 : PVP Nanofibers: a) Low Magnification (5% TiO_2), b) High Magnification (5% TiO_2), c) Low Magnification (10% TiO_2) d) High Magnification (10% TiO_2), e) Low Magnification (15% TiO_2) f) High Magnification (15% TiO_2)

50ml beaker and to it 2-methoxy ethanol solvent is added to make a homogenous solution. At 0.2 weight % of Ag the nanofibers had a diameter range of 145nm – 160nm. On increasing the weight % of silver to 0.4 the diameters of the nanofibers formed varied from 245nm to 260nm. However, on further increasing the silver concentration to 0.6 weight % and 0.8 weight % the diameters of the nanofibers varied approximately from 45nm to 170nm. The behavior of this change in diameter shrinkage is to be understood in detailed.

The nanofibers obtained were further tested for their antimicrobial properties for various bacterial and fungal species.

TiO₂: Zn Nanofibers

The fibers grown with Zn as dopant with 0.2 wt % concentration have been shown in the Fig. 7. Images of the fibers shows with different magnification from lower right to top left. As can be seen from the Fig. 4.1 with lower magnification, we could see the formation of the fibers on the Al foil substrate. With slight increase in magnification we could see the twisting behavior of the nanofibers with one another. With further close observation of the nanofibers and measurement of the diameter using the software we could see the diameter of the fiber is around 2.3 microns. However, in some area we could see the diameter of the fiber is as low as 650 nm as shown in the top left image in Fig. 7.

The fibers grown with Zn as dopant with 0.4 wt % concentration have been shown in the Fig. 8. Images of the fibers shows with different magnification from lower right to top left. As can be seen from the right side Fig. 8, with lower magnification, we could see the formation of the fibers on the Al foil substrate with almost high uniformity. With slight increase in magnification we could see the twisting behavior of the nanofibers with one another. With further close observation of the nanofibers and measurement of the diameter using the software we could see the diameter of the fiber is nearly 1.7 microns. However, in some area we could see the diameter of the fiber is as low as 230 nm as shown in the top left image in Fig. 8.

The fibers grown with Zn as dopant with 0.4 wt % concentration have been shown in the Fig. 9. Images of the fibers shows with different magnification from lower right to top left. As can

be seen from the Fig. 9 with lower magnification, we could see the formation of the fibers on the Al foil substrate with almost high uniformity. With slight increase in magnification we could see the twisting behavior of the nanofibers with one another and with two to three different diameters of the nanofibers. With further close observation of the nanofibers and measurement of the diameter using the software we could see the diameter of the fiber is around 1.9 microns. However, in some area we could see the diameter of the fiber is as low as 530 nm as shown in the top left image in Fig. 9.

The fibers grown with Zn as dopant with 0.4 wt % concentration have been shown in the Fig. 10. Images of the fibers shows with different magnification from lower right to top left. As can be seen from the Fig. 10 with lower magnification, we could see the formation of the fibers on the

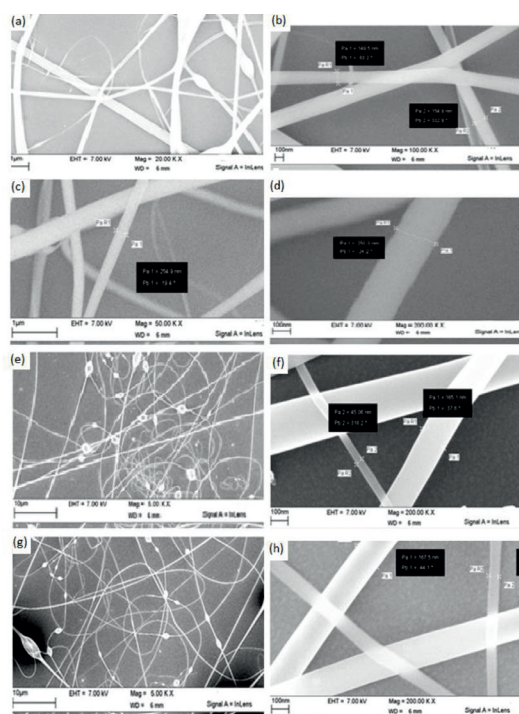


Fig. 6. SEM image of 15% PVP: 10% TiO₂: a) 0.2% Ag Nanofibers: Low Magnification, b) 0.2% Ag Nanofibers: High Magnification, c) 0.4% Ag Nanofibers: Low Magnification, d) High Magnification, e) 0.6% Ag Nanofibers: Low Magnification, f) 0.6% Ag Nanofibers: High Magnification, g) 0.8% Ag Nanofibers: Low Magnification, h) 0.8% Ag Nanofibers: High Magnification

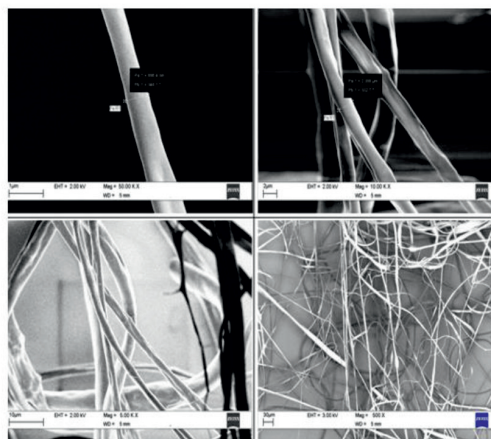


Fig. 7. SEM image of 15% PVP: 10% TiO₂: 0.2% Zn nanofibers

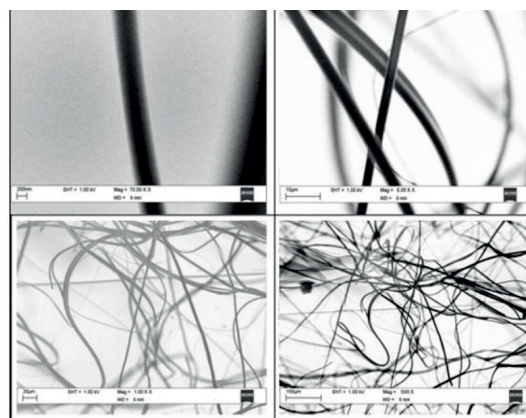


Fig 8: SEM image of 15% PVP: 10% TiO₂: 0.4% Zn nanofibers

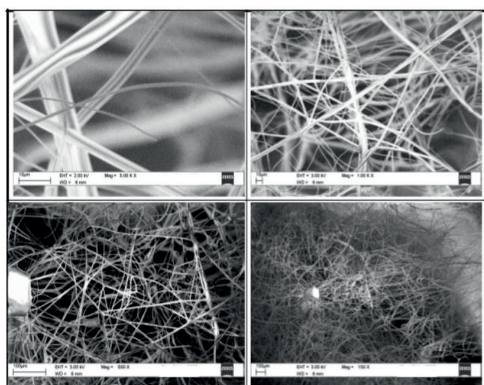


Fig. 9. SEM image of 15% PVP: 10% TiO₂: 0.6% Zn nanofibers

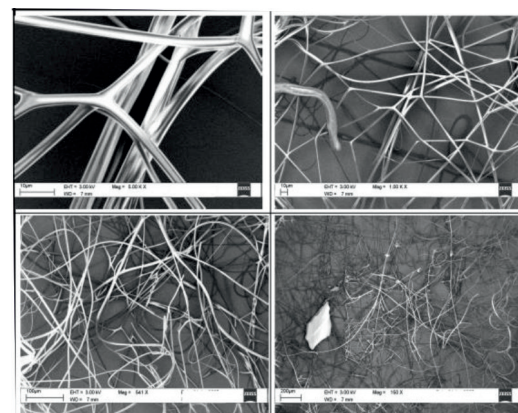


Fig. 10. SEM image of 15% PVP: 10% TiO₂: 0.8% Zn nanofibers

Al foil substrate with almost high uniformity. With slight increase in magnification we could see the twisting behavior of the nanofibers with one another and with two to three different diameters of the nanofibers. With further close observation of the nanofibers and measurement of the diameter using the software we could see the diameter of the fiber is around 2.1 microns. However, in some area we could see the diameter of the fiber is as low as 430 nm as shown in the top left image in Fig. 10.

CONCLUSION

The composition of PVP nanofibers were optimized at a concentration of 15 weight % in ethanol. The optimized composition was further used to synthesize TiO₂: PVP nanofibers.

Titanium dioxide: PVP nanofibers were synthesized at varying concentrations. TiO₂: PVP with 10wt % TiO₂ in ethanol was chosen as the optimized condition. TiO₂: PVP with 15wt % TiO₂ in ethanol was chosen as the optimized condition. Nanofibers of TiO₂: PVP doped with Ag were obtained using varying Ag concentrations. Nanofibers of TiO₂: PVP doped with Ag were obtained using varying Zn concentrations. The morphology of the obtained nanofibers was analyzed by Scanning Electron Microscopy (SEM), Fourier Transform Infra Red radiation (FTIR) and X-ray Diffraction (XRD). The anti bacterial testing was carried out on the following bacterial species; *Bacillus subtilis* (BS), *Staphylococcus epidermidis* (SE), *Staphylococcus aureus* (SA) and *Staphylococcus faecalis* (SF). The anti fungal testing was carried out on the

following species; *Candida tropicalis*(CT), *Candida parapsilosis* (CP), *Candida albicans* (CA) and *Candida glabrata* (CG).

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CONFLICT OF INTEREST

The authors declare that there is no conflict of interest.

AUTHORS' CONTRIBUTION

All authors have made substantial contribution to the work and approved it for publication.

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None.

DATA AVAILABILITY

All datasets generated or analyzed during this study are included in the manuscript.

ETHICS STATEMENT

This article does not contain any studies with human participants or animals performed by any of the authors.

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